

CAMOSUN COLLEGE School of Arts & Science Department of Chemistry & Geoscience

CHEM-220-001 Inorganic Chemistry Winter 2019

COURSE OUTLINE

The course description is online @ http://camosun.ca/learn/calendar/current/web/chem.html

 Ω Please note: This outline will <u>not</u> be kept indefinitely. It is recommended students keep this outline for their records, especially to assist in transfer credit to post-secondary institutions.

1. Instructor Information

(a) Instructor	John Lee	
(b) Office hours	See posted times on office door	
(c) Location	F 348 D	
(d) Phone	250 370 3436	Alternative:
(e) E-mail	leejohn@camosun.ca	
(f) Website	D2L	

2. Intended Learning Outcomes

Upon completion of this course the student will be able to:

- 1. Utilize a detailed knowledge of the electronic structure of atoms to rationalize many of the physical and chemical properties of atoms.
- Apply simple and sophisticated bonding theories to explain many of the properties of ionic and molecular substances.
- 3. Comment on the chemistry of the first row transition metals, especially in respect to formation of coordination compounds, their catalytic activity, and their relevance to bioinorganic chemistry.
- 4. Describe the major features of the chemistry of the main group elements of groups 14 and 18.
- 5. Use equipment associated with the preparation and analysis of inorganic compounds and perform reactions under an inert atmosphere for air- or water-sensitive compounds.
- 6. Outline the common approaches to synthesizing inorganic and coordination compounds in the laboratory.

3. Required Materials

(a) Texts

*Sapling Learning course code (~\$40 US)

Laboratory Manual. Inorganic Chemistry (Winter 2019), John Lee

*If you purchased the 2 course option in Fall, there is no need to buy another code.

(b) Other **OPTIONAL ONLY**: **INORGANIC CHEMISTRY**, Housecroft & Sharpe, *pub*: Pearson, current edition is 4ed. Other ed. acceptable.

(c)

4. Course Content and Schedule

Lecture Times: Mondays: 10:00 to 11:20 Y 214 Wednesday 2:00 to 3:20 Y 310 (TBC)

Lab Times: Friday 9:30 to 12:20 F 356

Note: This is only a preliminary lab schedule, changes can be made due to equipment &/or scheduling of other sections... Lab coat and eye protection are both mandatory and not provided!!

I	Week Number	Activity & Experiment Number	Actual Date of Lab
Jan 7th II	Begins on		<u>Friday</u>
II	I	Gold nanoparticles	Jan 11 th
Jan 14th III	Jan 7 th		
Jan 14th III	II	Quantum Dots	Jan 18 th
Jan 21st IV	Jan 14 th		
IV	III	Luminescent Lanthanides	Jan 25 th
Jan 28th V	Jan 21st	,	
Jan 28th V	IV	Isomerism- part I	Feb 1st
Feb 4th	Jan 28 th		
Feb 4th	V	Isomerism- part II	Feb 8th
Feb 11th VII [College Closed] Feb 22nd Feb 18th VIII [Titanium] Mar 1st Feb 25th IX [Optical activity at an octahedral cobalt centre] Mar 4th X [In(II) or tin (IV)? - part I Mar 15th XI [In(II) or tin (IV)? - part II Mar 22nd Mar 18th XII [Midterm 2 Mar 29th XIII Beer Cans to LED's part 1 Apr 5th XIV Element Jeopardy Apr 12th Final Exams	Feb 4th		
Feb 11th	VI	Midterm 1	Feb 15th
Feb 18th VIII [Titanium] Mar 1st Feb 25th IX [Optical activity at an octahedral cobalt centre] Mar 8th Mar 4th cobalt centre] X tin(II) or tin (IV)? - part I Mar 15th XI tin(II) or tin (IV)? - part II Mar 22nd Mar 18th XII Midterm 2 Mar 29th Mar 25th XIII Beer Cans to LED's part 1 Apr 5th XIV Element Jeopardy Apr 12th Final Exams	Feb 11th		
Feb 18th VIII Feb 25th IX Mar 4th Cobalt centre X Mar 11th XI Mar 18th XII Mar 18th XIII Mar 25th XIII Mar 25th XIII Apr 1st XIV Apr 8th Element Jeopardy Apr 12th Final Exams	VII	[College Closed]	Feb 22nd
Feb 25th IX [Optical activity at an octahedral Mar 8th Mar 4th cobalt centre Mar 15th X tin(II) or tin (IV) ? - part I Mar 15th XI tin(II) or tin (IV) ? - part II Mar 22nd Mar 18th Mar 25th Mar 25th XII Mar 25th Mar 25th XIII Beer Cans to LED's part 1 Apr 5th XIV Element Jeopardy Apr 12th Apr 8th Final Exams	Feb 18th		
Feb 25th IX [Optical activity at an octahedral Mar 8th Mar 4th cobalt centre Mar 15th X tin(II) or tin (IV) ? - part I Mar 15th XI tin(II) or tin (IV) ? - part II Mar 22nd Mar 18th Mar 25th Mar 25th XII Midterm 2 Mar 29th XIII Beer Cans to LED's part 1 Apr 5th XIV Element Jeopardy Apr 12th Apr 8th Final Exams	VIII	[Titanium]	Mar 1st
Mar 4th Cobalt centre	Feb 25th		
Mar 4th Cobalt centre	IX	Optical activity at an octahedral	Mar 8 th
Mar 11th	Mar 4 th		
Mar 11th XI	X	tin(II) or tin (IV) ? – part I	Mar 15th
Mar 18th XII Midterm 2 Mar 29th Mar 29th XIII Beer Cans to LED's part 1 Apr 5th Apr 1st XIV Apr 8th Final Exams	Mar 11 th		
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Final Exams Final Exams	Apr 8th	(Ľ
	•	Final Exams	
Apr 15 th to Apr 25 th	Final Exam Period	Apr 15th to Apr 23rd	

5. Basis of Student Assessment (Weighting)

(a) Assignments

Sapling Homework

Online assignments: 20 %

(b) Midterms

2 midterm tests, each worth 10 %

(c) Final Exam

Covers all course material: 35 %

(d) Laboratory Work

25 % includes reports and pre-labs

Chemistry is a practical science, it is essential to attend the laboratory classes. In the event of a student being unable to attend a laboratory class it is advised that the student attempt to obtain data from a partner or perform the class with another section in order to complete the assignment/report. It is mandatory that you give your Lab Instructor and/or Lab Partner the courtesy of an email in the event that you miss a laboratory class, otherwise a score of zero will be assigned. You must pass the lab component of the course to pass the course.

Students are responsible for obtaining their own safety glasses and laboratory jacket from the bookstore. It is not the responsibility of the College to provide you with safety equipment.

6. Grading System

X	Standard Grading System (GPA)
	Competency Based Grading System

7. Recommended Materials to Assist Students to Succeed Throughout the Course

Your instructor is always the best resource. In the event that office hours do not fit your schedule, please email for an appointment.

8. College Supports, Services and Policies



Immediate, Urgent, or Emergency Support

If you or someone you know requires immediate, urgent, or emergency support (e.g. illness, injury, thoughts of suicide, sexual assault, etc.), **SEEK HELP**. Resource contacts @ http://camosun.ca/about/mental-health/emergency.html or http://camosun.ca/services/sexual-violence/get-support.html#urgent

College Services

Camosun offers a variety of health and academic support services, including counselling, dental, disability resource centre, help centre, learning skills, sexual violence support & education, library, and writing centre. For more information on each of these services, visit the **STUDENT SERVICES** link on the College website at http://camosun.ca/

College Policies

Camosun strives to provide clear, transparent, and easily accessible policies that exemplify the college's commitment to life-changing learning. It is the student's responsibility to become familiar with the content of College policies. Policies are available on the College website at http://camosun.ca/about/policies/. Education and academic policies include, but are not limited to, Academic Progress, Admission, Course Withdrawals, Standards for Awarding Credentials, Involuntary Health and Safety Leave of Absence, Prior Learning Assessment, Medical/Compassionate Withdrawal, Sexual Violence and Misconduct, Student Ancillary Fees, Student Appeals, Student Conduct, and Student Penalties and Fines.

A. GRADING SYSTEMS http://camosun.ca/about/policies/index.html

The following two grading systems are used at Camosun College:

1. Standard Grading System (GPA)

Percentage	Grade	Description	Grade Point Equivalency
90-100	A+		9
85-89	Α		8
80-84	A-		7
77-79	B+		6
73-76	В		5
70-72	B-		4
65-69	C+		3
60-64	С		2
50-59	D		1
0-49	F	Minimum level has not been achieved.	0

2. Competency Based Grading System (Non GPA)

This grading system is based on satisfactory acquisition of defined skills or successful completion of the course learning outcomes

Grade	Description	
СОМ	The student has met the goals, criteria, or competencies established for this course, practicum or field placement.	
DST	The student has met and exceeded, above and beyond expectation, the goals, criteria, or competencies established for this course, practicum or field placement.	
NC	The student has not met the goals, criteria or competencies established for this course, practicum or field placement.	

B. Temporary Grades

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy at http://camosun.ca/about/policies/index.html for information on conversion to final grades, and for additional information on student record and transcript notations.

Temporary Grade	Description
I	Incomplete: A temporary grade assigned when the requirements of a course have not yet been completed due to hardship or extenuating circumstances, such as illness or death in the family.
IP	<i>In progress</i> : A temporary grade assigned for courses that are designed to have an anticipated enrollment that extends beyond one term. No more than two IP grades will be assigned for the same course.
CW	Compulsory Withdrawal: A temporary grade assigned by a Dean when an instructor, after documenting the prescriptive strategies applied and consulting with peers, deems that a student is unsafe to self or others and must be removed from the lab, practicum, worksite, or field placement.