



CAMOSUN COLLEGE
School of Arts & Science
Department of Chemistry & Geoscience

CHEM-120-003
College Chemistry 1
Fall 2018

COURSE OUTLINE

The course description is online @ <http://camosun.ca/learn/calendar/current/web/chem.html>

Ω Please note: This outline will not be kept indefinitely. It is recommended students keep this outline for their records, especially to assist in transfer credit to post-secondary institutions.

1. Instructor Information

(a) Instructor	John Lee
(b) Office hours	See posted times on office door and by appointment
(c) Location	F 348 D
(d) Phone	250 370 3436 Alternative: _____
(e) E-mail	leejohn@camosun.ca
(f) Website	D2L

2. Intended Learning Outcomes

Upon completion of this course the student will be able to:

1. Utilize nomenclature rules to name ionic and covalent compounds.
2. Demonstrate an understanding of stoichiometry by balancing chemical equations and performing mathematical calculations involving chemical reactions.
3. Describe the electronic structure of any atom in the periodic table and apply it to explain many of the physical and chemical properties of the elements.
4. Utilize simple bonding theories to explain why elements combine to form the compounds they do and also to explain many of the properties of compounds.
5. Apply knowledge of intermolecular interactions to rationalize many important physical properties of bulk matter in the gas, liquid and solid phases.
6. Use standard chemistry lab equipment, including burets, pipets, Buchner filters, and volumetric glassware in the correct manner.
7. Perform many standard laboratory procedures, such as titrations, preparation of standard solutions, the preparation, isolation, and purification of compounds, as well as use spectrophotometers to make analytical measurements.

3. Required Materials

(a) Texts

Mastering Chemistry Course Code. \$70. A Mastering Chemistry Access Code can be purchased from the bookstore. If you choose to purchase a new textbook or ebook (see below) then this includes a Mastering Chemistry Course Code. It is valid for 24 months after activation. If you have purchased the course code within the last 24 months it will still be valid for this course.

Other Recommended Materials for the Course

Chemistry, The Central Science, Brown, le May, Bursten. Custom Camosun Edition. \$145. (For hard copy and access code). ebook and access code \$110. Available from the Camosun Bookstore

The 2nd and 1st Australian editions are also acceptable if you have a used book. **If you buy a used book that is not the most recent edition you will still need to purchase a Mastering Chemistry access code.**

Chemistry 100, Camosun College course pack is a good source of review material for those students who may have been away from Chemistry for a while.

4. Course Content and Schedule

Lecture Times: Monday, Tuesday & Friday in F 210: All Lectures at 9.30 am - 10.20 am.

Laboratory classes: Thursday 9.30 to 12.20 pm F 356. Laboratory schedule and midterm dates are shown below.

Note: This is only a preliminary lab schedule, changes will be made due to equipment &/or scheduling of other sections... Lab coat and eye protection are both mandatory!!

Week Number	Activity & Experiment Number	Actual Date of Lab Thursdays
I Sep 3 rd	Lab Safety and Quiz: Attendance Mandatory unless previous lab credit has been granted	Sep 6 th
II Sep 10 th	Expt. 2 Densities of Solids & Liquids	Sep 13 th
III Sep 17 th	Group A Expt. 3 Stoichiometry of Chem. Rxns	Sep 20 th
IV Sep 24 th	Group B Expt. 3 Stoichiometry of Chem. Rxns	Sep 27 th
V Oct 1 st	Expt. 4 The Spectroscopic Determination of Nickel in Aqueous Solution	Oct 4 th
VI Oct 8 th	Midterm Test 1	Oct 11 th
VII Oct 15 th	Expt. 5 Colorimetric Determination of Iron in a Vitamin Tablet	Oct 18 th
VIII Oct 22 nd	Expt. 6 Determination of Copper Using Atomic Absorption Spectroscopy	Oct 25 th
IX Oct 29 th	Expt. 7 Hard Water determination	Nov 1 st
X Nov 5 th	Gas Laws Lab	Nov 8 th
XI Nov 12 th	Midterm Test 2	Nov 15 th
XII Nov 19 th	Expt. 9 The Preparation of Potassium Tris(oxalato)Ferrate(III)	Nov 22 nd
XIII Nov 26 th	Expt. 10 Analysis & Uses of Potassium Tris(oxalato)Ferrate(III)	Nov 29 th
XIV Dec 3 rd	Review	Dec 6 th
Final Exam Period	Final Exam TBD Dec 10 th to Dec 14 th	

5. Basis of Student Assessment (Weighting)

The course mark will be derived in the following manner:

Midterm tests	20 %
Final test	30 %
Laboratory work	25 %
Online Assignments	25 %

If it is advantageous to the student the theory mark will be solely derived from the final examination, or a combination of midterm with the final. **Online assignment marks may not be carried over so you must complete these before the due date. You usually have several weeks to do these assignments so no excuses will be accepted, no exceptions.**

In the event of the midterm test being missed due to illness/other commitments the weight of the missed test will be carried over to the final.

The Laboratory Mark

The breakdown of the Laboratory mark is as follows:

Pre-lab assignments (completed prior to starting the lab class).	12.5 %
Quality of Lab Performance/Reports/Assignments	87.5 %

No more than **2 laboratory classes may be missed**, during the course. In the event of a student being unable to attend a laboratory class it is advised that the student attempt to obtain data from a partner or perform the class with another section in order to complete the assignment/report. It is essential that you give your lab instructor the courtesy of an email in the event that you miss a laboratory class.

A student that attends the laboratory class but does not present a written report will receive a (maximum) score of 40%.

Students must receive a passing score of 50 % in the laboratory portion of the course, in order to be eligible to receive a passing grade in the course.

Students are responsible for obtaining their own safety glasses and laboratory jacket from the bookstore. It is not the responsibility of the College to provide you with safety equipment.

6. Grading System

Standard Grading System (GPA)

Competency Based Grading System

7. Recommended Materials to Assist Students to Succeed Throughout the Course

Your instructor is always the best resource. In the event that office hours do not fit your schedule, please email for an appointment.

The School of Arts and Science also offers science help centres. The chemistry help centre operates out of Y302. The hours are posted on the website: <http://camosun.ca/learn/subjects/chemistry/> and on the notice board along the hallway of the chemistry laboratories.

8. College Supports, Services and Policies



Immediate, Urgent, or Emergency Support

If you or someone you know requires immediate, urgent, or emergency support (e.g. illness, injury, thoughts of suicide, sexual assault, etc.), **SEEK HELP**. Resource contacts @ <http://camosun.ca/about/mental-health/emergency.html> or <http://camosun.ca/services/sexual-violence/get-support.html#urgent>

College Services

Camosun offers a variety of health and academic support services, including counselling, dental, disability resource centre, help centre, learning skills, sexual violence support & education, library, and writing centre. For more information on each of these services, visit the **STUDENT SERVICES** link on the College website at <http://camosun.ca/>

College Policies

Camosun strives to provide clear, transparent, and easily accessible policies that exemplify the college's commitment to life-changing learning. It is the student's responsibility to become familiar with the content of College policies. Policies are available on the College website at <http://camosun.ca/about/policies/>. Education and academic policies include, but are not limited to, Academic Progress, Admission, Course Withdrawals, Standards for Awarding Credentials, Involuntary Health and Safety Leave of Absence, Prior Learning Assessment, Medical/Compassionate Withdrawal, Sexual Violence and Misconduct, Student Ancillary Fees, Student Appeals, Student Conduct, and Student Penalties and Fines.

A. GRADING SYSTEMS <http://camosun.ca/about/policies/index.html>

The following two grading systems are used at Camosun College:

1. Standard Grading System (GPA)

Percentage	Grade	Description	Grade Point Equivalency
90-100	A+		9
85-89	A		8
80-84	A-		7
77-79	B+		6
73-76	B		5
70-72	B-		4
65-69	C+		3
60-64	C		2
50-59	D		1
0-49	F	Minimum level has not been achieved.	0

2. Competency Based Grading System (Non GPA)

This grading system is based on satisfactory acquisition of defined skills or successful completion of the course learning outcomes

Grade	Description
COM	The student has met the goals, criteria, or competencies established for this course, practicum or field placement.
DST	The student has met and exceeded, above and beyond expectation, the goals, criteria, or competencies established for this course, practicum or field placement.
NC	The student has not met the goals, criteria or competencies established for this course, practicum or field placement.

B. Temporary Grades

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy at <http://camosun.ca/about/policies/index.html> for information on conversion to final grades, and for additional information on student record and transcript notations.

Temporary Grade	Description
I	<i>Incomplete:</i> A temporary grade assigned when the requirements of a course have not yet been completed due to hardship or extenuating circumstances, such as illness or death in the family.
IP	<i>In progress:</i> A temporary grade assigned for courses that are designed to have an anticipated enrollment that extends beyond one term. No more than two IP grades will be assigned for the same course.
CW	<i>Compulsory Withdrawal:</i> A temporary grade assigned by a Dean when an instructor, after documenting the prescriptive strategies applied and consulting with peers, deems that a student is unsafe to self or others and must be removed from the lab, practicum, worksite, or field placement.