School of Arts & Science



DEPARTMENT OF CHEMISTRY & GEOSCIENCE

Chemistry 255 BIOCHEMISTRY

Winter Semester 2011

COURSE OUTLINE

This course concerns fundamental aspects of biomolecules and biochemical processes. Topics include: noncovalent interactions; buffers; protein, enzyme, and carbohydrate structure-function relationships; lipids and membranes; bioenergetics; carbohydrate, lipid, amino acid and nucleotide metabolism; nucleic acid structure and synthesis; gene expression and protein synthesis; nutrition; biotechnology applications; and prevalent biochemistry laboratory techniques.

The Approved Course Description is available on the web @ <u>http://www.camosun.bc.ca/calendar/chem.php#255</u>

Please note: This outline will not be kept indefinitely. It is recommended students keep it for their records.

1. Instructor Information

- (a) Instructor Jamie Doran, Ph.D.
- (b) Office hours Monday, 1:30 to 2:20 pm Tuesday, 11:30 to 1:20 pm Wednesday, 1:30 to 2:20 pm Thursday, 1:30 to 2:20 pm Friday, 1:30 to 2:20 pm

Everyone is welcome whenever my office door is open. Appointments may be made to meet at times other than those listed above. Office hours will be extended prior to exams.

- (c) Location Room 342C, Fisher Building, Lansdowne Campus
- (d) Phone 250.370.3441
- (e) E-mail jdoran@camosun.bc.ca

2. Intended Learning Outcomes

Students successful in this course will be able to:

- Demonstrate an understanding of the fundamental characteristics of proteins, including enzymes, carbohydrates, lipids, and the nucleic acids, DNA and RNA.
- Demonstrate a basic understanding of the chemical and biochemical principles governing the structure-function relationships of biomolecules and membranes.
- Demonstrate an understanding of the underlying themes of key biochemical processes including metabolism, bioenergetics, gene expression and protein synthesis.
- Demonstrate an understanding of, and evaluate, the most important aspects of the biochemical framework of cellular function at a molecular level, and the chemical bases thereof.
- Use the basic vocabulary of biochemistry.
- Use internet-based resources to enhance the learning and appreciation of the principles of biochemistry.
- Demonstrate an appreciation of the significance of biochemistry in clinical and veterinary medicine, laboratory analyses, nutrition, agriculture, and biotechnology.
- Isolate specific proteins using gel-permeation, ion-exchange, and affinity-based column chromatography methods.
- Analyze proteins by SDS-polyacrylamide gel electrophoresis (SDS-PAGE).
- Conduct polymerase chain reaction amplification of DNA molecules.
- Utilize agarose gel electrophoresis for analysis of DNA samples.
- Critically analyze the results obtained using each of the biochemical experimental techniques described above.
- Work with an awareness of the basic safety considerations and general procedures associated with a biochemistry laboratory.

3. Course Materials

(a) **Text:**

Principles of Biochemistry. Fourth Edition. 2006. H.R. Horton, L.A. Moran, K.G. Scrimgeour, M.D. Perry & J.D. Rawn. Prentice-Hall Canada Inc., Toronto.

This textbook is required for this course.

Links to relevant web-based learning resources are presented in the text. A copy of the textbook is available in the reserve library of the Lansdowne Campus. **Copies of the most relevant lecture slides are contained in one of the course packs available through the Lansdowne book store.** This course package is **very highly recommended**.

(b) Laboratory Manual & Study Guides:

Chem 255 *Experimental Procedures & Course Study Guide.* 2010-2011 *Edition* This is a course pack that contains the Laboratory Manual as well as supportive materials for experimental procedures. Importantly, this also contains a collection of study guides for all subjects covered in each chapter of the textbook and other materials included in the lecture portion of the course. *This course package is required material* and is available through the Camosun College Lansdowne campus Book Store.

(c) Chem 255 Lecture Slides package, 2010-2011 Edition.

The experience of past students for many years has indicated that this package is extremely beneficial to note-taking, and to promoting lecture-based learning and inclass discussion, and to learning in general. *Therefore, it is very highly recommended.* It is available through the Camosun College Lansdowne campus book store.

(d) General Materials and Supplies

- <u>Safety glasses</u> Safety glasses are required when handling hazardous chemicals or biochemicals, or glassware. <u>Each student is required to provide her or</u> <u>his pair of safety glasses</u>. Students lacking safety glasses when they are required will not be permitted to work in the laboratory.
- <u>Lab coats</u> Lab coats are required for all experimental work in the laboratory. <u>Each</u> <u>student is required to provide her or his own lab coat.</u> Students lacking lab coats will not be permitted to work in the laboratory.
- <u>Latex gloves</u> Latex or other 'non-allergenic' gloves *will be available in the lab* and are to be used when appropriate to protect the skin from potentially hazardous chemicals or, more frequently, to protect valuable biochemicals from contamination or becoming degraded by enzymes from the skin.
- <u>Scientific calculator</u> Calculators may be required in the lab, in class and during exams. Each student is required to provide her or his own calculator.

4. Course Content and Schedule

| Credits | 4 credits |
|---------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| In-class workload | 6 hours per week There are four 50-minute lectures per week. Term exam review periods will be scheduled into an appropriate lecture slot prior to each exam. Experiments, pre-lab talks & post-lab analyses are conducted during most of the one hour & 50 minute Friday time slots. This time period is also used for the two term exams, and for a final exam review period. |
| Out-of-class workload | 6 hours per week |
| Number of weeks | 14 weeks |
| Pre-requisite Pre- or Co-requisite | Chem 121 - College Chemistry 2 Chem 230 – Organic Chemistry 1 |

Course times and locations

| <u>Lectures</u> | Monday, 10:30 AM - 11:20 AM Fisher Building, Room F360 |
|---------------------|------------------------------------------------------------------------------------------|
| | Tuesday, 10:30 AM - 11:20 AM Fisher Building, Room F360 |
| | Wednesday, 10:30 AM - 11:20 AM Fisher Building, Room F216 |
| | Thursday, 10:30 AM - 11:20 AM Fisher Building, Room F360 |
| Lab Experiments (or | <u>Term Tests*)</u> Friday 10:30 AM - 12:20 PM Fisher Building, Rooms F360/F358 |
| | |

* Please see the laboratory and term test schedule below.

Lecture Outline

A general outline of the topics to be covered in the course is provided below. *Study* guides for each chapter of the textbook are provided in the 'course package' available through the College Bookstore. Each study guide includes an assigned reading list for the chapter, a listing of the relevant vocabulary, and suggested practice questions.

 Introduction to Biochemistry
 Chapter 1

 Introduction; History; Physiologically Relevant Elements, Classes of Organic
 Compounds, Functional Groups; Covalent Linkages; Classes of Biomolecules. (Other review material in this chapter forms part of the assigned reading listed in the study guide)

Noncovalent bonding, pH, pKa, and Buffers & Buffering Chapter 2 Noncovalent Bonding in Biomolecules; pKa; The Henderson-Hasselbach Equation; Buffering; the Bicarbonate Blood-Buffer System; Acidosis & Alkalosis (additional material not presented in Chapter 2). (Other review material in this chapter forms part of the assigned reading)

Amino Acids and the Primary Sequence of a Protein

Structures of Common Amino Acids; Ionization & pKa's of Amino Acid Functional Groups; Peptide Bonds; Protein Purification Techniques; Primary Protein Sequence; Protein Sequencing; Comparative Analyses of Protein Sequences; MALDI-TOF.

Protein Structure & Function

Informatics - Proteomics, Genomics, Systeomics, & Metabonomics (*additional material not presented in Chapter 4 but covered in part in Chapter 23);* The Nature of the Peptide Bond; Secondary, Tertiary and Quaternary Protein Structures; Protein Folding and Stability; Protein Structure-Function Relationships.

Enzymes

Classes of Enzymes; Enzyme Kinetics; Michaelis-Menton Equation; Enzyme Inhibition; Interpretation of Lineweaver-Burk Plots; Regulation of Enzyme Activity.

Mechanisms of Enzyme Catalysis

Overview of Enzyme Function; Mechanisms of Enzyme Catalysis; Mechanism of Chymotrypsin Activity.

Coenzymes and Vitamins

Vitamins and Health (overview); Nature, Roles and Structures of Coenzymes and Vitamins.

Carbohydrates

Roles and Structures of Monosaccharides, Disaccharides, Polysaccharides & Glycoconjugates.

Lipids & Membranes

Classes and Structures of Lipids, Membrane Structures, Membrane Transport, Transmembrane Signal Transduction.

Chapter 4

Chapter 3

Chapter 6

Chapter 5

Chapter 7

Chapter 8

Chapter 9

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| Overview of Metabolism Introduction to Intermediary Metabolism and Bioenergetics. | Chapter 10 | | | |
|---------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|--|--|
| Glycolysis The Metabolic Pathway of Glycolysis and its Regulation. | Chapter 11 | | | |
| TCA Cycle Mitochondrial Transport of Pyruvate; Pyruvate Dehydrogenase Regulation; The Citric Acid Cycle (Krebs) Cycle; Regulation of Kreb's Cy | | | | |
| Other Pathways in Carbohydrate Metabolism Glycogen Metabolism; Gluconeogenesis; Cori Cycle; Pentose Phos Maintenance and Regulation of Blood Glucose Levels. | <i>Chapter 13</i> phate Pathway; | | | |
| Electron Transport and Oxidative PhosphorylationChapter 14Introduction to Bioenergetics; The Chemiosmotic Hypothesis; Electron Transport;Oxidative Phosphorylation in Mitochondria; Malate-Aspartate Shuttle System. | | | | |
| Lipid MetabolismChapter 16Dietary Fats and Health (Overview); Lipoprotein Structure and Function; Storage and Mobilization of Fatty Acids and Cholesterol; Fatty Acid β-Oxidation; Ketone Bodies; Fatty Acid, Phospholipid and Cholesterol Metabolism. | | | | |
| Amino Acid Metabolism Nitrogen Assimilation; Introduction to Amino Acid Metabolism; Urea C | Chapter 17 | | | |
| Witogen Assimilation, introduction to Aminto Acta Metabolism, orea e | cycle. | | | |
| Nucleotide Metabolism Introduction to Nucleotide Metabolic Pathways. | Cycle. <i>Chapter 18</i> | | | |
| Nucleotide Metabolism | <i>Chapter 18</i> <i>Chapter 19</i> ases; Restriction | | | |
| Nucleotide Metabolism Introduction to Nucleotide Metabolic Pathways. DNA Composition, Structure and Mapping Introduction to Nucleic Acids; Nucleotides & Nucleosides; Nucleotide | <i>Chapter 18</i> <i>Chapter 19</i> ases; Restriction | | | |
| Nucleotide Metabolism Introduction to Nucleotide Metabolic Pathways. DNA Composition, Structure and Mapping Introduction to Nucleic Acids; Nucleotides & Nucleosides; Nuclea Endonucleases and Physical Mapping of DNA; DNA Finger-Printing (all DNA Replication and Repair | <i>Chapter 18</i> <i>Chapter 19</i> ases; Restriction so see Ch. 23). <i>Chapter 20</i> <i>Chapter 21</i> | | | |
| Nucleotide Metabolism Introduction to Nucleotide Metabolic Pathways. DNA Composition, Structure and Mapping Introduction to Nucleic Acids; Nucleotides & Nucleosides; Nucleat Endonucleases and Physical Mapping of DNA; DNA Finger-Printing (all DNA Replication and Repair DNA Polymerase; DNA Replication; DNA Sequencing; DNA Repair. RNA Synthesis (Transcription) Classes of RNA; RNA Polymerase Function & Promoter Sequences; Transcription | <i>Chapter 18</i> <i>Chapter 19</i> ases; Restriction so see Ch. 23). <i>Chapter 20</i> <i>Chapter 21</i> ascriptional <i>Chapter 22</i> <i>Chapter 22</i> <i>Chapter 22</i> <i>Chapter 22</i> | | | |

Recombinant DNA Technologies & BiotechnologyChapter 23Basic Molecular Cloning Strategies; PCR (also see lab info); Site-Directed Mutagenesis.

Laboratory, Exam & Assignment Schedule

<u>Friday, January 14th.</u> Overview of a general strategy for protein purification and characterization, and the relationships between the inter-related experiments.

<u>Friday, January 21st.</u> Experiment 1 Separation of Proteins by Gel Permeation Column Chromatography

<u>Friday, January 28th.</u> Experiments 2 & 3 Ion-Exchange Column Chromatography & Affinity Column Chromatography

<u>Friday, February 4th.</u> Experiment 3 & Data Analyses from Expts. 1, 2 & 3 Separation of Proteins by Affinity Chromatography (continuation). Consideration of Column Chromatography Techniques.

<u>Friday, February 11th.</u> Experiment 4, Part I SDS-Polyacrylamide Gel Electrophoresis (SDS-PAGE): Separation & Identification of Proteins, and Determination of Protein Molecular Weight Part 1 - Preparation of a Polyacrylamide Gel for the Separation of Proteins

Friday, February 18th. Term Test 1 - 10:30 am to 12:20 pm, Rm. F360

Friday February 25th. Reading Break

<u>Friday, March 4th.</u> Experiment 4 - Part 2 SDS-Polyacrylamide Gel Electrophoresis (SDS-PAGE) Separation & Identification of Proteins, and Determination of Protein Molecular Weight Part 2 - Polyacrylamide Gel Electrophoresis, & Staining for Detection of Proteins

<u>Friday, March 11th.</u> Experiment 4 - Part 3 SDS-Polyacrylamide Gel Electrophoresis (SDS-PAGE) Separation & Identification of Proteins, and Determination of Protein Molecular Weight Part 3 –Analysis of SDS-PAGE Results

<u>Friday, March 18th.</u> Experiment 5 - Part 1 Partial Analysis using the Polymerase Chain Reaction (PCR) of Cloned SAGE Tag Fragments Representing Genes Expressed by Pre-Cancerous Cells Part 1 - PCR Amplification of Cloned SAGE Tag DNA Fragments.

<u>Friday, March 25th.</u> Experiment 5 - Part 2 Partial Analysis using the Polymerase Chain Reaction (PCR) of Cloned SAGE Tag Fragments Representing Genes Expressed by Pre-Cancerous Cells Part 2 - Agarose Gel Electrophoresis & Detection of PCR Amplified DNA Fragments.

April March 1st Term Test 2 - 10:30 am to 12:20 pm, Rm. F360

<u>Friday, April 8th.</u> Experiment 5 - Part 3 Partial Analysis using the Polymerase Chain Reaction (PCR) of Cloned SAGE Tag Fragments Representing Genes Expressed by Pre-Cancerous Cells Part 3 - Analysis of PCR Results.

Experiment 6 - Analyses of Triglycerides, Total Cholesterol, HDL and LDL Levels. <u>Friday, April 15th</u>. Final Exam Review Semester Project Due

<u>Final Exam</u>: The time and location of the Chem 255 Final Exam will be published by the College during the Winter Semester.

5. Basis of Student Assessment (Weighting)

(a) Assignment: <u>Metabolic Pathways Chart Project.</u>

This assignment will be described in detail in a handout to be provided once topics directly concerning intermediary metabolism arise in the course. This is an out-of-class project. Each individual is required to hand in the results of her or his own work. This metabolic pathways chart is due on the final day of class, but may be handed in earlier. It should be considered very useful for study purposes in preparation for the final exam. The value this project contributes to the final grade is **5%**.

(b) Laboratory Experiments

Attendance in the lab periods is mandatory. No laboratory experiment can be missed without an acceptable reason submitted in writing such as a proper letter from a Medical Doctor. *Please come to each lab period prepared for each experiment*. There are no laboratory reports due for the experiments but *students are responsible for understanding the principles, technical bases, and results of each experiment*. These aspects of the laboratory work may be subject to examination on the term exams and the final exam.

(c) Term Tests

Term Test #1

This exam covers relevant material from approximately the first third of the course. The delineation of material students are responsible for will be provided in class about one week before the date of the test. This is a 110 minute test that will be written on Friday, February 18th from 10:30 AM to 12:20 PM in F360. The results of this test contribute to **25%** of the final grade.

Term Test #2

This exam covers relevant material from approximately the second third of the course. The delineation of material students are responsible for on this test will be provided in class about one week before the date of the exam. This is a 110 min. test that will be written on <u>Friday</u>, <u>April 1st</u> in room F360 from 10:30 AM to 12:20 PM. The results of this test contribute to **25%** of the final grade.

If either of the term exams is missed due to illness or other justifiable reason (appropriate documentation required), a student may either take a substitute test to be written at a mutually agreeable time, or choose to add the percentage value of that midterm exam (25%) to the percentage value of the final exam.

(d) Final Exam

The final exam is a comprehensive exam.

The value this exam contributes to the final grade is 45%.

Attendance at the final exam is mandatory. Appropriate documentation must accompany any explanation for absence. The time and location of the final exam will be published by the College during the semester.

6. Grading System

Standard Grading System (GPA)

| Percentage | Grade | Description | Grade Point Equivalency |
|------------|-------|--------------------------------------|----------------------------|
| 90-100 | A+ | | 9 |
| 85-89 | А | | 8 |
| 80-84 | A- | | 7 |
| 77-79 | B+ | | 6 |
| 73-76 | В | | 5 |
| 70-72 | B- | | 4 |
| 65-69 | C+ | | 3 |
| 60-64 | С | | 2 |
| 50-59 | D | | 1 |
| 0-49 | F | Minimum level has not been achieved. | 0 |

Temporary Grades

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy at **camosun.ca** for information on conversion to final grades, and for additional information on student record and transcript notations.

| Temporary Grade | Description |
|--------------------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| I | <i>Incomplete</i> : A temporary grade assigned when the requirements of a course have not yet been completed due to hardship or extenuating circumstances, such as illness or death in the family. |
| IP | <i>In progress</i> : A temporary grade assigned for courses that are designed to have an anticipated enrollment that extends beyond one term. No more than two IP grades will be assigned for the same course. |
| CW | <i>Compulsory Withdrawal:</i> A temporary grade assigned by a Dean when an instructor, after documenting the prescriptive strategies applied and consulting with peers, deems that a student is unsafe to self or others and must be removed from the lab, practicum, worksite, or field placement. |

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy

E-1.5 at **camosun.ca** for information on conversion to final grades, and for additional information on student record and transcript notations.

7. Recommended Materials or Services to Assist Students to Succeed Throughout the Course

The chapter, laboratory and course study guides provided in the course package, which also includes the experimental protocols, will prove very valuable. In addition to the practice problems provided, there are lists of representative problems from the text and the corresponding website. Also, the textbook includes additional problems and their answers, and provides links to websites. Together these resources will further enhance the understanding and appreciation of the curriculum of this biochemistry course.

LEARNING SUPPORT AND SERVICES FOR STUDENTS

There are a variety of services available for students to assist them throughout their learning. This information is available in the College Calendar, Registrar's Office or the College web site at http://www.camosun.bc.ca

Please Note:

Students may not use recording devices in the classroom without the prior permission of the instructor. However, the instructor's permission is not required when the use of a recording device is sanctioned by the College's Resource Centre for Students with Disabilities in order to accommodate a student's disability and when the instructor has been provided with an instructor notification letter which specifies the use of a recording device. Recordings made in the classroom are for the approved student's personal use only, and distribution of recorded material is prohibited.

ACADEMIC CONDUCT POLICY

There is an Academic Conduct Policy. It is the student's responsibility to become familiar with the content of this policy. The policy is available in each School Administration Office, Registration, and on the College web site in the Policy Section.

www.camosun.bc.ca/divisions/pres/policy/2-education/2-5.html