



School of Arts & Science
CHEMISTRY AND GEOSCIENCE DEPARTMENT

CHEM 121-01
College Chemistry 2
2008P

COURSE OUTLINE

The Approved Course Description is available on the web @ camosun.bc.ca

Ω Please note: this outline will be electronically stored for five (5) years only.
It is strongly recommended students keep this outline for your records.

1. Instructor Information

(a)	Instructor:	Diana Li	
(b)	Office Hours:	Tue 2:30-4:30; Wed—before &/or after class, depending on Di's availability... Additional office hours posted on Di's door.	
(c)	Location:	F344C	
(d)	Phone:	370-3444	Alternative Phone: try the Chemistry Hotline!
(e)	Email:	lid@camosun.bc.ca	
(f)	Website:	http://camosun.ca/learn/programs/chem.html	

2. Intended Learning Outcomes

(No changes are to be made to this section, unless the Approved Course Description has been forwarded through EDCO for approval.)

Upon completion of this course the student will be able to:

1. Utilize the specialized vocabulary and nomenclature based on the IUPAC system of organic compounds to name and draw structures for many simple organic compounds containing the common functional groups.
2. Write chemical reactions to illustrate numerous transformations between organic functional groups.
3. Draw structural and stereoisomers of organic compounds and name stereoisomers based upon the IUPAC system of nomenclature.
4. Demonstrate an understanding of the factors that influence the rate of a chemical reaction, deduce the rate of a chemical reaction from time/concentration data, and utilize rate laws to perform kinetic calculations.
5. Apply the laws of thermodynamics and account for the factors that lead to spontaneous physical and chemical changes.
6. Explain how and why reactions attain equilibrium positions and perform calculations pertaining to equilibrium systems.
7. Describe redox reactions, use electrochemical data to predict the spontaneity of redox reactions, and comprehend the structures of electrochemical cells.
8. Describe various acid-base theories and apply these theories to acid-base reactions in aqueous solution.
9. Perform experiments in the areas of preparative organic, preparative inorganic, physical and analytical chemistry and use the various associated pieces of laboratory equipment.

3. Required Materials

(a)	Texts	"Chemistry, The Central Science: a broad perspective" by Brown et. al., 2007—a.k.a. B-L-B Australian Edition.
(b)	Other	♦ Chem 121 Lecture Notes Supplement & Appendix by D. Li, 2008 Ed. This supplement goes with both the 10 th & Australian editions of B-L-B although the Australian edition is preferred. Note: homework questions selected from both editions are not necessarily identical but equivalent. ♦ Chem 121 Lab Manual. ♦ Darling Molecular Model—highly recommended if you are taking 2 nd year Chemistry courses (Chem 230, 231, etc.). ♦ Safety glasses mandatory & lab coat recommended

4. Course Content and Schedule

(Can include: class hours, lab hours, out of class requirements and/or dates for quizzes, exams, lectures, labs, seminars, practicums, etc.)

Lecture Plan:		
Chapter in Aust. Ed.	(10 th Ed.)	Topic (approximate number of lecture hours)
21, 22, 23—selected topics		Organic Chemistry (9)
12	(14)	Chemical Kinetics (7)
4	(5)	Thermochemistry (4)
17	(19)	Thermodynamics (3)
13	(15)	Chemical Equilibrium (3)
14	(16)	Acid-Base Equilibria (4)
15	(17)	Additional Aspects of Aqueous Equilibria (5)
18	(20)	Electrochemistry (3)

5. Basis of Student Assessment (Weighting)

(Should be linked directly to learning outcomes.)

Labs (8 or 9 experiments)	22%
Test I (Organic Chemistry)	18% (Week III, May 22, up to 2½ hours)*
Test II (Kinetics & Thermochemistry)	20% (Week V, June 4, up to 2½ hours)*
Final Exam (comprehensive)	40% (TBA ~ Week IV, 3 hours in June)

* Test dates to be confirmed during the first week of classes in May.

Notes:

- (1) Student must pass the lab portion of the course to obtain credit for Chem 121.
- (2) Student is encouraged to attempt both tests. Test scores that is not as high as that of the June final exam will be dropped automatically and its weight redistributed to the final exam. You may choose not to write one or both tests, your final exam will then be 78% of the course grade!!
- (3) Student must write each test as scheduled. No one is allowed to write late and there will be no exceptions. Early exam is a privilege and not a right; thus, at full discretion of the instructor.

6. Grading System

(No changes are to be made to this section, unless the Approved Course Description has been forwarded through EDCO for approval.)

Standard Grading System (GPA)

Percentage	Grade	Description	Grade Point Equivalency
90-100	A+		9
85-89	A		8
80-84	A-		7
77-79	B+		6
73-76	B		5
70-72	B-		4
65-69	C+		3
60-64	C		2
50-59	D	Minimum level of achievement for which credit is granted; a course with a "D" grade cannot be used as a prerequisite.	1
0-49	F	Minimum level has not been achieved.	0

Temporary Grades

Temporary grades are assigned for specific circumstances and will convert to a final grade according to the grading scheme being used in the course. See Grading Policy E-1.5 at camosun.ca for information on conversion to final grades, and for additional information on student record and transcript notations.

Temporary Grade	Description
I	<i>Incomplete:</i> A temporary grade assigned when the requirements of a course have not yet been completed due to hardship or extenuating circumstances, such as illness or death in the family.
IP	<i>In progress:</i> A temporary grade assigned for courses that, due to design may require a further enrollment in the same course. No more than two IP grades will be assigned for the same course. (For these courses a final grade will be assigned to either the 3 rd course attempt or at the point of course completion.)
CW	<i>Compulsory Withdrawal:</i> A temporary grade assigned by a Dean when an instructor, after documenting the prescriptive strategies applied and consulting with peers, deems that a student is unsafe to self or others and must be removed from the lab, practicum, worksite, or field placement.

7. Important Dates

Week

II May 12 (Mon): Fee Deadline

III May 19 (Mon): Victoria Day

Test I: May 22 (Thu), 11:30-14:00 in E201 (location to be confirmed)

V June 4 (Wed): Last Day to Withdraw or Change to Audit...

Test II: June 4 (Wed), 11:30-14:00 in E201

June 23-25: Exam Period for Spring 2008

8. Recommended Materials or Services to Assist Students to Succeed Throughout the Course

Articles in the Library Reserve Room for Chem 120 & 121:
(at least one copy of the followings)

9th & 10th Editions B-L-B: Text & Solutions Manual, Student's Guide, "Math Review Toolkit" & "The Essentials of Organic Chemistry" booklet. (Note: short answers to all red numbered exercises are at the back of the Text!)

Australian B-L-B (2 copies) &

♦ OneKey—a text-specific online resource tool: www.pearsoned.com.au/brown.

(Note: Short answers to all red numbered exercises are at the back of the Text and all answers to red & black numbered exercises are available—email Di or photocopy hard copies on Di's door!)

LEARNING SUPPORT AND SERVICES FOR STUDENTS

There are a variety of services available for students to assist them throughout their learning. This information is available in the College calendar, at Student Services or the College web site at camosun.ca.

STUDENT CONDUCT POLICY

There is a Student Conduct Policy **which includes plagiarism**. It is the student's responsibility to become familiar with the content of this policy. The policy is available in each School Administration Office, at Student Services and on the College web site in the Policy Section.

ADDITIONAL COMMENTS AS APPROPRIATE OR AS REQUIRED